Carboxamides and Hydrazide of Glycopeptide Antibiotic Eremomycin

Synthesis and Antibacterial activity

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Carboxamides and hydrazide of glycopeptide antibiotic eremomycin were obtained by a direct phosphorazidate as a condencing agent. Eremomycin hydrazide was also obtained by hydrazinolysis of the eremomycin methyl ester. Use of dicyclohexylcarbodiimide or 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide for amidation led to the corresponding eremomycin ureides. The ESI-MS data indicate that eremomycin and its amides exist as dimers. The carboxamide, methylamide and benzylamide of eremomycin were as active against Gram-positive bacteria as the parent antibiotic, and the methylamide, benzylamide and hydrazide were almost an order of magnitude more active than eremomycin against Staphylococcus epidermidis clinical isolates in vitro. Amide of eremomycin as well as ureides were devoid of histamine liberating properties, which demonstrates that protection of the carboxyl group leads to a decrease in the allergenic properties. that protection of the carboxyl group leads to a decrease in the allergenic properties.

Antibiotic eremomycin (1) is a potent antibacterial antibiotic of the dalbaheptide group closely related to vancomycin¹⁾. The molecular basis for the antibacterial activity of the group arises from the specific binding of activity of the group arises from the specific binding of σ gives proposed to the nascent bacterial cell wall bearing the C-terminal sequence D-Ala-D-Ala. Although the affinity of these antibiotics for the bacterial cell wall analogues in some cases correlates correlates directly with the interest of th *vitro* antibiotic activity, eremomycin, which is more active than vancomycin against a range of staphylococcal strains, has a lower affinity for the cell wall analogue di-N-acetyl-L-Lys-D-Ala-D-Ala. Recently it was demondi-7V-acetyl-L-Lys-D-Ala-D-Ala. Recently it was demonstrated that eremomychi forms stable dimers $(\mathbf{N}_{\text{dim}})$ 3×10^6 , for vancomycin K_{dim} 700) and that the dimeriza-
tion plays a role in the mechanism of action of these antibiotics^{2,3)}. This peculiarity of eremomycin makes antibiotics². This peculiarity of eremomycin makes investigation of structure-activity relationships among extends both of properties both of properties both of properties and theoretical and theoretical and theoretical and theoretical and the properties of properties and the properties of properties and the properties of prope interest.

The model of the eremomycin dimer formation and interaction with the D-Ala-D-Ala moiety suggests that the carboxyl group of the antibiotic participates neither the carboxyl group of the antibiotic participates neither in the dimerization nor in the interaction with the target and thus affords an opportunity for valuable chemical modifications. Eremomycin analogs modified at the

carboxyl group may retain the antibacterial properties, whereas the modification may favorably influence other whereas the modification may favorably influence other pharmacological properties, first of all, allergen Methyl ester of eremomycin (2) demonstrated high

with or even superior to that or eventomycin $\frac{1}{\sqrt{2}}$ for glycope and its deglycosylated derivatives, transformation into carboxamides led in some cases to compounds with antibacterial activity superior to the starting antibiotic, or specie interest is some in vitro activity of the carboxamides against VanA enterococci highly resistant to both teicoplanin and vancomycin⁶⁾. In the synthesis of carboxamides of teicoplanine and its deglycosylated products, diphenyl phosphorazidate (DPPA) as a condensing agent or activated cyanomethyl eaters were used. For the synthesis of amides of the glycopeptide antibiotics ardacin, parvodicin or its derivatives, the \overline{M} and its derivatives, the intervals of \overline{M} TV-protected glycopeptides were condensed with the $\frac{1}{\sqrt{2}}$ corresponding to presence of also presence of alkylchloroformates³. Recently, two methods for the synthesis of vancomycin carboxamide derivatives by using DCC or HBTU (2-(1-hydroxybenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate) were reported^{8,9)}.

 ζ Carboxamides and hydrazide of eremomycin were obtained by a direct reaction of the carboxyl group of as a condencing agent; if the NH_2 -compound was used as a salt, the reaction was carried out in the presence of $Et₃N$. Preliminary protection of amino groups was not necessary in this case. Using this method eremomycin $\frac{1}{1}$ in this case. Using the case of this method eremoment $\frac{1}{1}$ carboxamide (4), hydrazide (5), methylamide (6), $\begin{bmatrix} 0 & 1 \\ 0 & 1 \end{bmatrix}$, 1 $\begin{bmatrix} 0 & 1 \\ 0 & 1 \end{bmatrix}$, and morphologyperation (8), and more probably percent (9) were synthesized. Hydrazide 5 was also obtained by the hydrazinolysis of ester $2⁴$.

Use of dicyclohexylcarbodiimide for the amidation led to 1,3-dicyclohexylureide 10. With the use of l-(3 dimethylaminopropyl)-3-ethylcarbodiimide, a mixture α is defined ureful and α was isolated (α : 1. 1 as $\frac{1}{2}$ shown by HPLC). Isolation of the carboxamides hydrazide and ureides $4 \sim 12$ was performed by column
chromatography on CM-cellulose. The modification in the eremomycin molecule was localized in the peptide. the eremomycin increase was localized in the peptide molety after acid hydrolysis (yielding unmodif

eremosamines) and Edman's degradation. For the analysis of elemoniyem derivatives electromore straightforward and fast than FAB method⁵⁾. ESI-MS is capable of providing both molecular weight and structural information of this class of antibiotic and, due to its soft nature, prevents artifacts resulting f rom thermal degradation and/or from the ionization

process itself. In the ESI-MS spectrum of the eremomycin standard, in addition to the base peak $(m/z 1557.6)$, lowest standard, in addition to the base peak (m/z \sim 1557.6, lowest peak (m/z \sim 1557.6, lowest peak (m/z 1557.6, lowest pe isotope), which corresponds to the monoproton In particular, the peaks at m/z 1039.4 and m/z 774.4 In particular, the peaks at m/z 1039.4 and m/z 774.4 correspond to the triply charged triprotonated dimer $([2M + 3H]^{+3})$ and to the doubly charged diprotonated
monomer $([M + 2H]^{+2})$, respectively. The peaks at m/z 1485.5, 1413.5, 991.4, and 707.3 correspond to the loss of the eremosamine moiety (144 Da) from the doubly charged diprotonated dimer ($\lceil 2M + 2H - \text{erem} \rceil^{2+}$), from the monoprotonated monomer ($[M + H - \text{erem}]^+$), from the triply charged triprotonated dimon $(5M + 2H)$ the triply charged triprotonated dimer $(L2M + 3H)$ erem] $3+$) and from the doubly charged diiprotonated monomer ($[M+2H-$ erem^{$]^{2+}$}), respectively. All the investigated eremomycin amides investigated exhibited the same mass spectrometric behavior. Results of the ESI-MS investigation give an independent evidence of existence of eremomycin derivatives in dimeric forms. It is noteworthy that modifications of the carboxyl group do not prevent dimerization (at least, as it follows from $\mathbf{H}_{\mathbf{c}}$ not \mathbf{M} at $\mathbf{d}_{\mathbf{a}}$ is follows from \mathbf{R} the ESI-MS data).

Antibacterial properties of the compounds obtained were investigated *in vitro* and *in vivo*. Derivatives $4, 6$, and 7 were as active as the parent antibiotic against and 7 were as active as the parent antibiotic against Gram-positive bacteria. It should be noted that the carboxamides 6 and 7 and hydrazide 5 are almost by an order of magnitude more active than eremomycin against the Staphylococcus epidermidis clinical isolates (Table 1). This type of antibacterial activity is of special interest as $T_{\rm F}$, the antibacterial activity is of special interest as pathogenicity of S. epidermidis is associated with indwelling foreign devices, prosthetic valve endocarditis, surgical wound infections and is especially typical in immunosupressed patients. The compounds investigated
did not inhibit growth of the *Enterococcus faecalis* clinical isolates resistant to vancomycin (MIC > 128~mag/ml). An \ddot{x} is the contraction of definition (M) is \ddot{x} and \ddot{x} is \ddot{x} \boldsymbol{u} vitro investigation of defivatives 10 and $\boldsymbol{\Pi}$ +12 in comparison with eremomycin and vancomycin showed that the ureides are less active than 1 and comparable with vancomycin (for S. aureus, $MIC₅₀$ values for 10 and 11+12 are 4 and 2 μ g/ml, respectively; for eremo-
mycin and vancomycin, 0.5 and 2 μ g/ml, respectively).

A study of the antibacterial activity of amides in mice $\frac{1}{2}$ study of the antibacterial activity of amides in microscopic in microscopic in microscopic in microscopic in $\frac{1}{2}$ with acute ictifal staphylococcal infection *in vivo* demonstrated that carboxamide 4 and ester 3 when given
sc are close to the parent antibiotic whereas methylamide 6, is two-fold less active. Hydrazide 5 is four-fold less active than eremomycin and close to vancomycin (Table 2) when administered iv, amide 4 was tree-fold less active \overline{a} when administered in a was tree-fold less active \overline{a} than eremomycin $(ED_{50}$ 1.25 and 0.44 mg/kg, respectively

Table 1. Antibacterial evaluation of eremomycin amides and hydrazide in comparison with eremomycin (1); MIC values (μ g/ml).

Strain	Compound						
		4	5	6	7	8	9
<i>Staphylococcus aureus</i> Tour	0.13	0.13	0.25	0.25	0.13	0.5	0.5
<i>S. aureus</i> Tour 30% bovine serum	0.25	0.5	0.25	0.25	0.5		$\overline{2}$
<i>S. aureus</i> Smith	0.13	0.13	0.5	0.13	0.13	0.5	0.5
<i>S. aureus</i> clinical isolate	0.5	0.5		0.5	0.25		\overline{c}
S. epidermidis ATCC 12228	0.25	0.25	0.13	0.25	0.25	0.25	0.25
S. epidermidis clinical isolate		0.25	0.13	0.13	0.13	0.25	0.25
S. haemoliticus clinical isolate	0.25	0.13	0.13	0.5	0.13		0.25
Streptococcus pyogenes C203	0.13	0.13	0.13	0.13	0.06	0.13	0.13
S. pneumoniae UC41	0.13	0.13	0.13	0.13	0.13	0.13	0.13
Enterococcus faecalis ATCC 7080	0.25	0.25	0.5	0.25	0.13	0.25	0.13

tively).

Allergenic properties of the antibiotics of the vancomycin group represent the most important side- $\frac{1}{\sqrt{2}}$ is the most inportant $\frac{1}{\sqrt{2}}$ in the most important side $\frac{1}{\sqrt{2}}$ in the most important side of the most in the mos effect of these compounds. They depend on the endogenous histamine liberation which induces in rats This test has been previously used for determining the his tamine liberation by various substances¹⁰⁾. The lesions were measured after intraperitoneal administration of 1 and its derivatives 2, 3, 4, 10, $11 + 12$ in doses of 150 and $\frac{1}{2}$ 30 and 30 does not the LDS (and 30 does). Carboxamide α 4 as well as ureides 10, $11+12$ were devoid of the histamine liberating properties; esters 2 and 3 in this test were very weakly allergenic. It demonstrates that the protection of the carboxyl group leads to a decrease in $p \sim 1$ the aneigenicity (Table β).

Experimental

Eremomycin sulfate (1) was produced at the pilot plant of Institute of New Antibiotics. NMR spectra were recorded on a Varian VXR-400 spectrometer in D_2O at 70° C. Paper electrophoresis was performed in 0.05 AcOH-pyridine buffer (pH 5.6) at $900\,\text{V}$ for 3 hours on $A \in \mathbb{R}^n$ by \mathbb{R}^n $\{0, \ldots, \ldots, \ldots, \mathbb{R}^n\}$ the Filtrak F_N-12 paper (Germany). CM-cellulos

(CM-32 Whatman) column chromatography was per-
formed with LKB Ultragrad Gradient Mixer 11300
supplied with Uvicord 2138 and Recorder 2065. TLC was performed on the Merck Silica Gel $60F_{254}$ plates in $\frac{1}{100}$ expresses the Merck Silica Gel $\frac{1}{254}$ plates in the systems EtOAC-PrOH-25% NH4OH 2:2: 1 (A) $2:2:5$ (B), and $3:3:4$ (C). The samples were analyzed
by loop injection electrospray mass spectrometry using the conditions listed below: HPLC: Instrument Waters 625 LC System; eluent 0.1% aqueous TFA-CH₃CN 70: 30; flow rate 50 μ l/minute. ESI mass spectra were obtained using a Finnigan TSQ700 instrument equipped obtained using a Finnigan TSQ700 instrument equipped with a Finnigan Electrospray fon source; capillary pressure 70 psi, auxiliary gas (N_2) flow 20 ml/minute; scan range/rate $300 \sim 1800$ m/3 secesonds. In order to obtain range/rate 300~ 1800m/3 sececonds. In order to obtain maximun sensitivity and accuracy, the instrument was previously tuned and calibrated using a mixture of MRFA-myoglobin. Prior to analysis, specific tune the eremomycin standard (ca. 4mg/ml). About 0.5mg of each sample was dissolved in 250μ of mobile phase immediately before the analysis. immediately before the analysis.

 \overline{DPPA} (0.043 ml, 0.2 mmol) was added to a stirre $22 \text{ mg } (0.4 \text{ mmol})$ of NH Cl and 0.07 m (0.5 mmol) 22 mg (0.4mmol) of M_{4} c) and 0.07ml (0.5mmol) of

Et₃N in 8 ml dimethylsulfoxide. The reaction mixture was stirred for about 8 hours at 20 $^{\circ}$ C, and then another portion of DPPA (0.043 ml) were added. The stirring was continued for 14 hours at 20° C, then 150 ml of acetone was added, and the precipitate was collected, dissolved in 5 ml of 0.2M CH₃COONH₄ (pH 6.7) and applied to a chromatographic column with CM-cellulose $(1 \times 45 \text{ cm})$ preequilibrated with 0.2M CH_3 COONH₄ (pH 6.7). The preequilibrated with 0.2m CH3COONH4(pH 6.7). The column chromatograpy was carried out with 0.2m CH₃COONH₄ with a linear pH gradient (pH 6.7 \rightarrow 9.5).
The fractions containing 4 were acidified with 6 N H₂SO₄ to pH 6 and desalted by stirring with XAD -resin (50 ml) for 24 hours followed by elution with 150 ml of $H₂O$ - $CH₃OH$ (1:1) mixture. The eluate was concentrated to a small volume (about 2 ml) in vacuo at 50 $^{\circ}$ C and 100 ml of acetone was added. The precipitate formed was collected, washed with acetone and dried in vacuo at 20° C overnight to give $39 \text{ mg } (23\%)$ of 4. Rf 0.14 (A); 0.62 (B), 0.42 (C). $E_{1 \text{cm}}^{1\%}$ 40 (280 nm). $[\alpha]_{D}^{20} - 100^{\circ}$ (c 1, water). ESI-MS: 1555 [M]^+ , C₇₃H₉₀N₁₁O₂₅Cl, Calc. 1555.

 $\frac{2.266 \text{ mJ/mol}}{A}$. 5 was obtained from 1 and NH₂NH₂ HCl

B. Hydrazine hydrate (0.04 ml, 0.6 mmol) was added to a stirred solution of 83 mg (0.05 mmol) of 2 in 5 ml of methanol. After 5 hours at 37° C, the formed precipitate was filtered off, washed with methanol and dried in vacuo. The purification was performed as described for 4 to give $29 \text{ mg } (17\%)$ of 5. Rf 0.64 (B); 0.43 (C). $E_{1 \text{ cm}}^{1\%}$ 40 (280) nm). $[\alpha]_D^{20}$ -98° (c 1, water). ESI-MS: 1570 [M]⁺, $C_{73}H_{91}N_{12}O_{25}Cl$, Calc. 1570.

 $\frac{\text{Eremon form new (c)}}{\text{DDDA}}$ (0.042 mL 0.2 mm sl) DPPA $(0.043 \text{ mi}, 0.2 \text{ mi})$ was added to a stirre solution of 165 mg (0.1 mmol) of 1 in 8 ml of DMSO and 0.05 ml of 30% aqueous CH_3NH_2 . The reaction mixture was stirred for 8 hours at 20° C, and another portion (0.043 ml) of DPPA and 0.1 ml of 30% aqueous CH₃NH₂ were added. After 14 hours of stirring, additional 0.086 ml of DPPA was added, and in 20 hours the product was of DPPAwas added, and in 20 hours the product was precipitated with 150ml of acetone. Purification was performed as described for 4 to yield 33 mg (20%) of 6.
Comparison of the ¹H NMR spectra of 6 and eremomycin demonstrates the presence of a six-proton singlet at 2.86 ppm, which corresponds to two methyl groups (N -terminal and methylamide ones). Rf 0.18 (A); 0.68 (B); 0.48 (C). $E_{1 \text{cm}}^{1\%}$ 41 (280 nm). $[\alpha]_{D}^{20}$ –98° (c 1, water). ESI-MS: 1569 [M]⁺, C₇₄H₉₂N₁₁O₂₅Cl, Calc. 1569.

 $\frac{1}{\text{DPPA}}$ (0.043 ml, 0.2 mmol) was added to a stirre solution of 165mg (0.1mmol) of 1 and 0.043ml (0.4 mmol) of benzyl amine in 8 ml of DMSO. The reaction mixture was stirred for 8 hours at 20° C, and two portions (0.086 ml each) of DPPA were added with a 14 hours interwal. The stirring was continued for 14 $\frac{1}{2}$ and $\frac{150 \text{ m}}{2}$ was edded and the precipitate hours, acetone (150 ml) was added and the precipitations,

was purified as described for amide 4 to give 35 mg of 7 (20%). In the ¹H NMR spectrum, five-proton multiplet in low field at $7.6 \sim 7.3$ ppm, corresponding to phenyl protons, is present. Rf 0.47 (A); 0.66 (C); $E_{1 \text{ cm}}^{1\%}$ 38 (280 nm). $[\alpha]_D^{20} - 96^\circ$ (c 1, water). ESI-MS: 1645 [M]⁺, $C_{80}H_{96}N_{11}O_{25}Cl$, Calc. 1645.

Eremomycin 4-Methylpiperazide (8)
It was obtained as described above in 18% yield. In ¹H NMR spectrum of 8 three-proton singlet of N-Me group and signals of four CH₂ groups at $4.10 \sim 3.37$ and $3.15 \sim 2.90$ ppm are present. Rf 0.23 (A); 0.57 (C). $E_{1cm}^{1\%}$ 3.40 (280 nm). $[\alpha]_D^{20} -101^\circ$ (c 1, water). ESI-MS: 1638 $[M]^+$, C₇₈H₉₉N₁₂O₂₅Cl, Calc. 1638.

Eremomycin Morpholide (9)
It was obtained as described above in 17% yield. Rf 0.25 (A); 0.59 (C). $E_{1 \text{ cm}}^{1\%}$ 39 (280 nm). $[\alpha]_{D}^{20}$ -98° (c 1, water). In the ¹H NMR spectrum of 9, signals of four
CH₂ groups at $3.68 \sim 3.21$ ppm are present. ESI-MS: 1625 [M]⁺, C₇₇H₉₆N₁₁O₂₆Cl, Calc. 1625.

 \overline{DCC} (21 mg, 0.1 mmol) was added to a solution of 165 mg (0.1 mmol) of 1 in 8 ml of DMSO at 20° C, after 8 hours addition of 21 mg of DCC was repeated. Dicyclohexylurea was filtered off after 14 hours, and the reaction product was precipitated with ethyl acetate and purified as described above to give 54 mg (29%) of 10. Rf 0.50 (A), 0.31 (B). $E_{1 \text{ cm}}^{1\%}$ 38 (280 nm). $[\alpha]_{D}^{20}$ –93° (c 1, methanol). In the ¹H NMR spectrum of 10 signals of methanol). In the π H nMR spectrum of 10 signals of the dicyclohexylureido group are present. Esi-MS: 1762 $[M]^+$, C₈₆H₁₁₁N₁₂O₂₆Cl, Calc. 1762.

A Mixture of Eremomycin 1-Ethyl-3-(3-dimethylami-
nopropyl)ureide (11) and Eremomycin 3-Ethyl-1-(3-di $rac{1}{\text{math-lominomallyuroida}}$ (17) methylaminopropyl)ureide (12)

It was obtained from 165 mg (0.1 mmol) of 1 and 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (76 mg) as described above in a yield of 112 mg (60%). Rf 0.41 (A); 0.20 (B). $E_{1 \text{cm}}^{1\%}$ 40 (280 nm). $[\alpha]_{D}^{20}$ –95° (c 1, water). In the ¹H NMR spectrum of $11+12$, signals of the In the $\frac{11 \text{ NML}}{2 \text{ times the probability of the number of times.}}$ Af-ethyl-Af'-(3-dimethylaminopropyl)ureido group are present. ESI-MS: 1711 $[M]^+$, $C_{81}H_{106}N_{13}O_{26}Cl$, Calc. 1711.

$\frac{1}{\sqrt{2}}$

Experiments In Vitro
Microtiter method; overnight incubation 37° C; inocula were at a final dilution of $1/500$ of the overnight broth culture. Todd-Hewitt broth was used for all enterococci culture. Todd-Hewitt broth-was used for all enterococci and strep cocci, nutrient broth was used for all other all other all other all other all other all other all o cultures.

Experiments *In Vivo*
Therapeutical effects of the amides in comparison with eremomycin and its N , N -dimethyl methyl ester (3) were eremomycin and its A^TV-dimethyl methyl ester (3) were

determined against an acute lethal staphylococcal in-
fection in albino mice $(BALB/c, 18+2 g)$ challenged by intraperitoneal injection with bacterial cells suspended in east is music with bacterial contracted central in gastric mucin sufficient to kill nontreated control within $24 \sim 48$ hours. The test compounds were administrated sc 30 minutes post infection and 24 hours later. In each test five mice were treated per one dose level. On the 7th day ED_{50} were calculated on the basis δ parameters of surviving miss of each does \mathbf{r} percentage of surviving microscopers \mathbf{r}

Histamine Liberating Properties

Singular intraperitoneal injection of the compounds in the dose of 150 and 300 mg/kg (0.05 and 0.1 of LD_{50} for eremomycin) were given to rats. Each group consisted of $6~10$ white rats weighting $110~130g$ each. The $\frac{1}{2}$ for $\frac{1}{2}$ here $\frac{1}{2}$ is $\frac{1}{2}$ and $\frac{1}{2}$ and $\frac{1}{2}$ are $\frac{1}{2}$ and $\frac{1}{2}$ are $\frac{1}{2}$ a tion of the drugs. Intensity of changing of stomach mucosa was macroscopically marked in four ball system and ulcers were marked with four balls if they covered $50 \sim 85\%$ of stomach mucosa. Three balls correspond to 5O ~85%of stomach mucosa. Three balls correspond to changes of $30 \approx 30\%$ of surface and two balls to $10~30\%$, smaller changes were marked with one ball. (Table 3).

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